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Same reason! Our acetylide ion is a strong base, and deprotonates the O-H group, which then participates in an S<sub>N</sub>2 reaction with the alkyl halide 4 bonds away (forming a five membered ring).. This is a textbook example of what we saw in our last post - an intramolecular SN2 reaction. [Why doesn't it do substitution first? Acid-base reactions are fast, relative to substitution reactions].

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